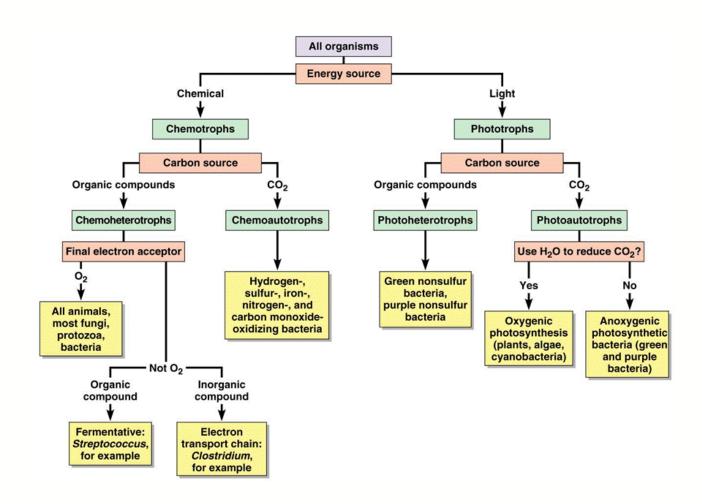
#### Early Metabolisms: Energetics

#### Sarah Maurer

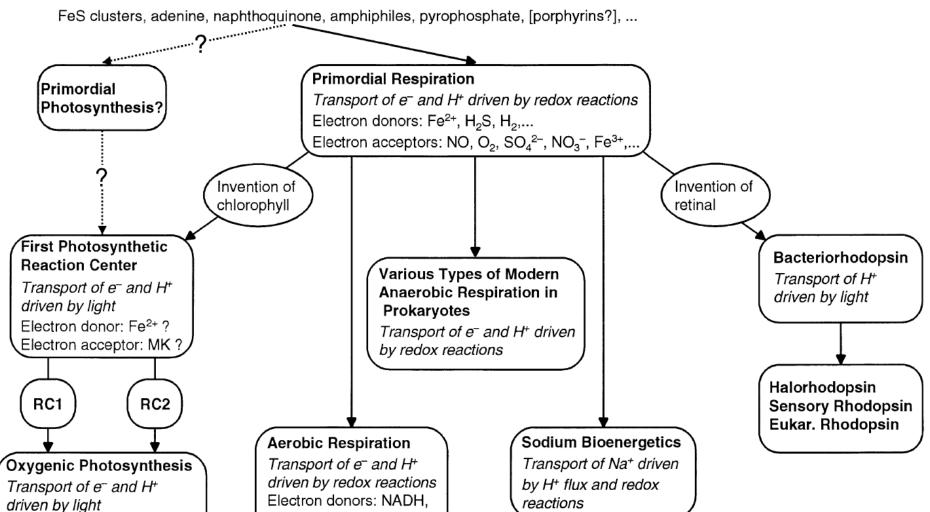
Central Connecticut State University Department of Chemistry and Biochemistry



## Classification of energy generation in organisms



## **Evolution of Metabolism**

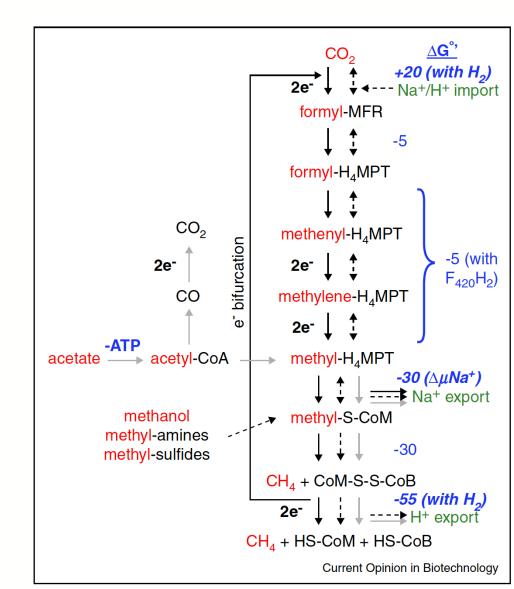


succinate (mitochondria)

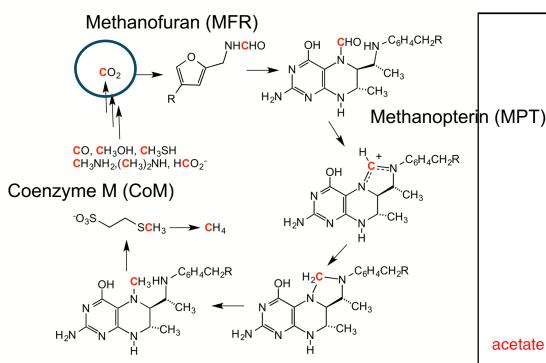
Electron acceptor: O2

Electron donor: H<sub>2</sub>O

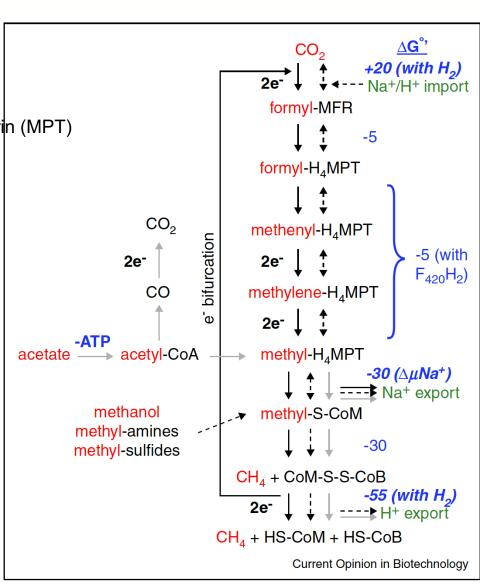
Electron acceptor: NADP

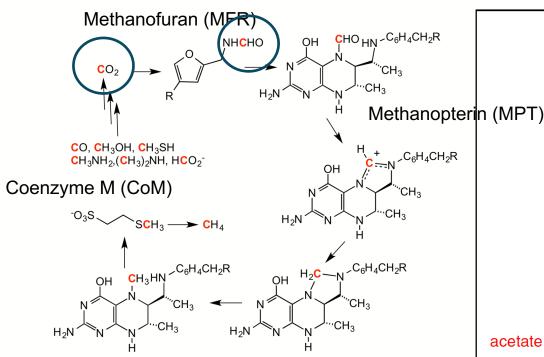


Costa, K. C.; Leigh, J. A. Metabolic Versatility in Methanogens. *Curr. Opin. Biotechnol.* **2014**, 29 (1), 70–75.

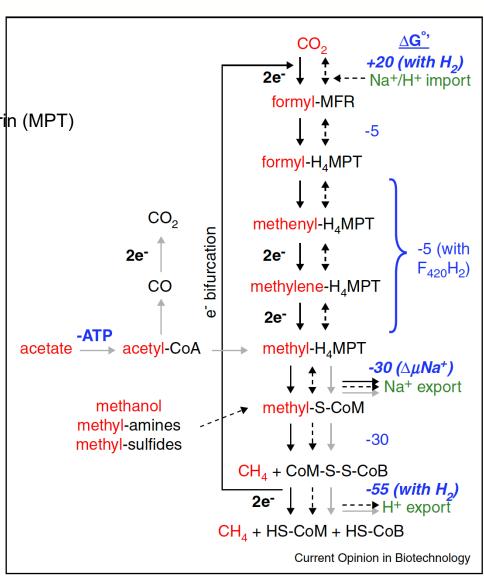


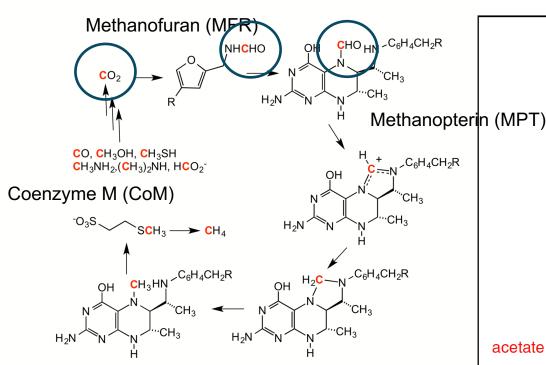
Costa, K. C.; Leigh, J. A. Metabolic Versatility in Methanogens. *Curr. Opin. Biotechnol.* **2014**, 29 (1), 70–75.



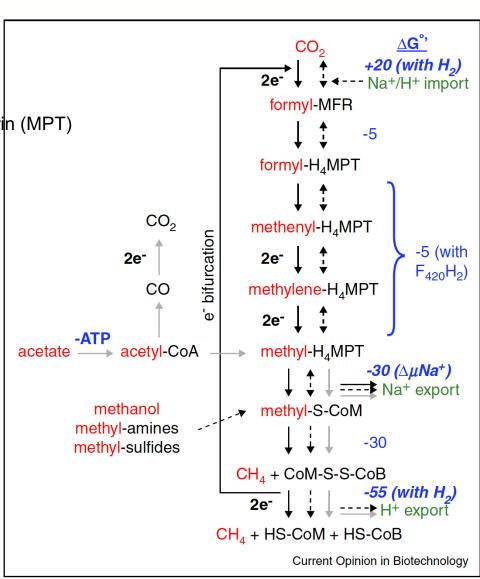


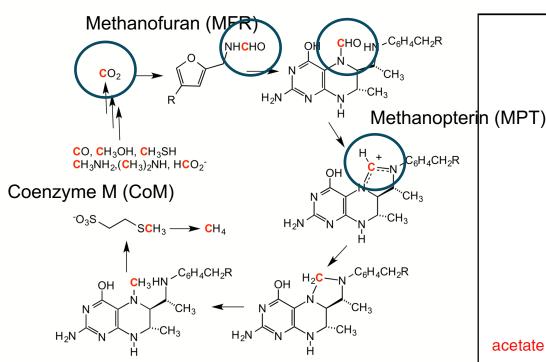
Costa, K. C.; Leigh, J. A. Metabolic Versatility in Methanogens. *Curr. Opin. Biotechnol.* **2014**, 29 (1), 70–75.



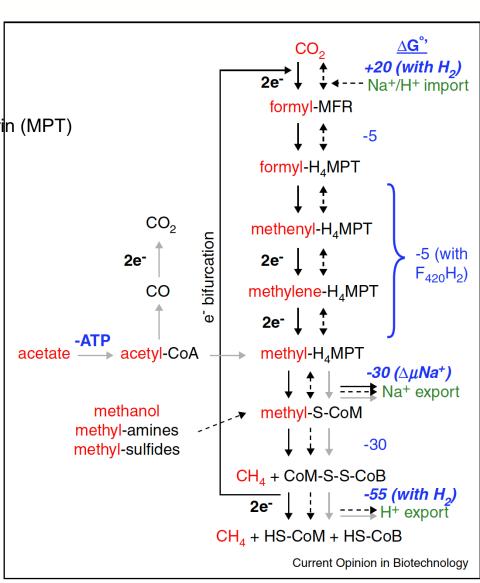


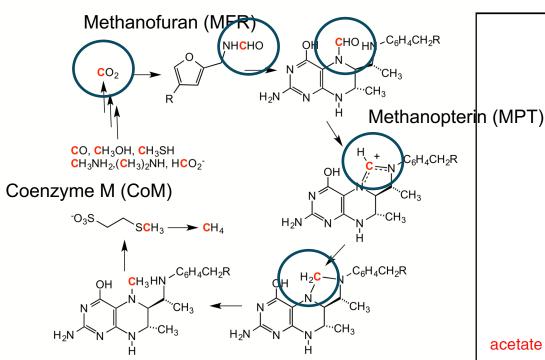
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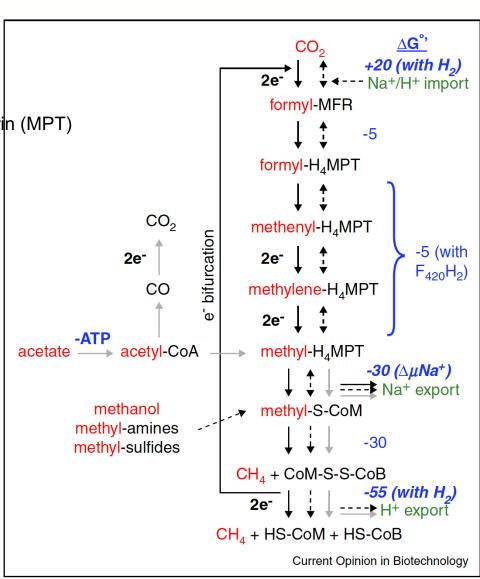


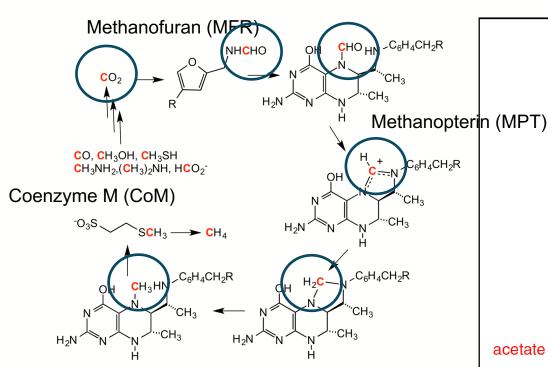
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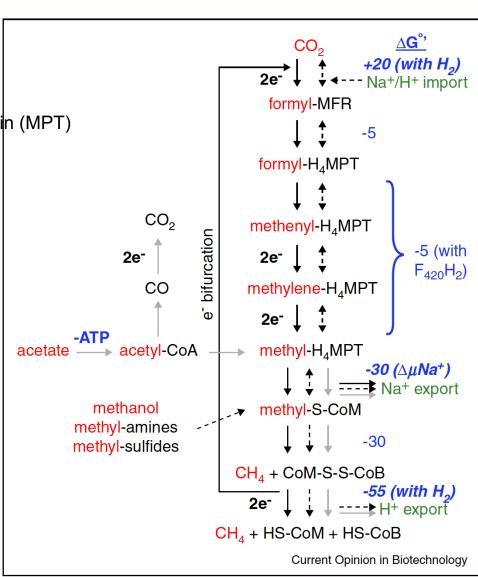


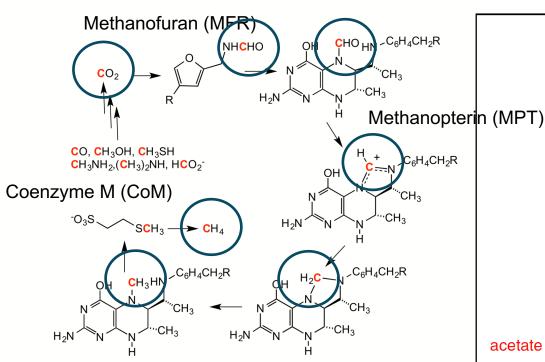
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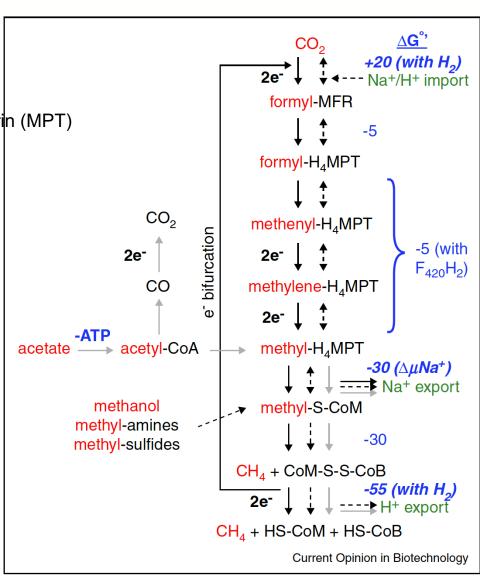


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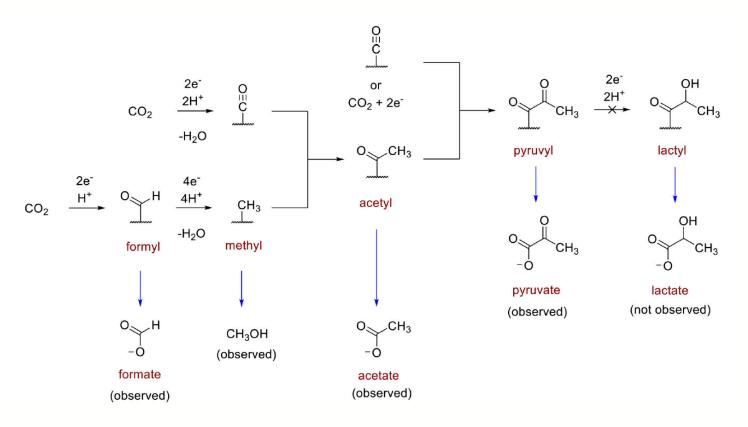




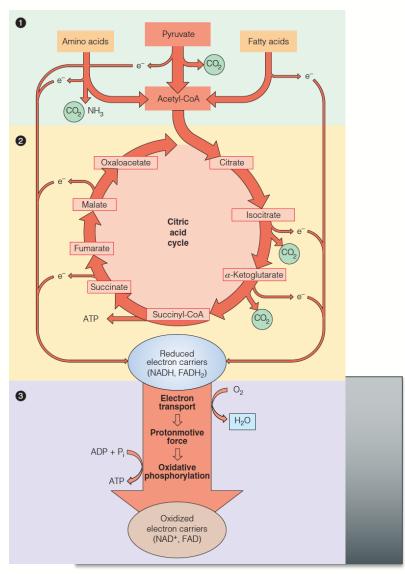
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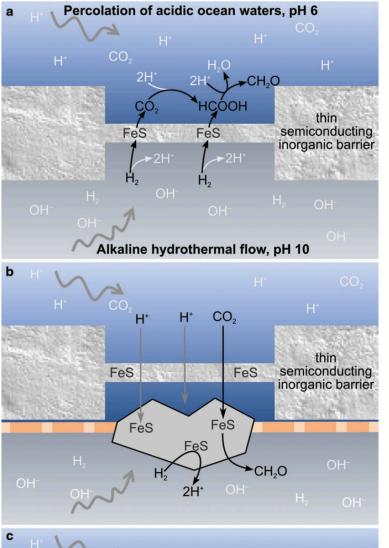


## Carbon fixation on minerals



#### Modern Redox Chemistry





CO.

FeS

# Primordial Redox Chemistry?

$$\begin{array}{ll} \text{H}_2 \rightarrow 2\text{H}^+ + 2\text{e}^- & \text{Loses electrons oxidized,} \\ \text{FeS} + \text{H}_2\text{S} \rightarrow \text{FeS}_2 + \text{H}_2 & \text{(LEO GER)} \\ \\ 4\text{H}_2 + \text{CO}_2 \rightarrow \text{CH}_4 + 2\text{H}_2\text{O} \\ \\ 4\text{H}_2 + 2\text{HCO}_3^- + \text{H}^+ \rightarrow \text{CH}_3\text{COO}^- + 2\text{H}_2\text{O} \\ \\ \text{H}_2 + \text{Fd}_{\text{ox}} \rightarrow \text{Fd}_{\text{red}}^{2-} + 2\text{H}^+ \end{array}$$

FeS

CO, H

